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FLUIDIZATION XVI

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Gaseous emissions during oxy-fuel combustion of sewage sludge in a circulating fluidized bed

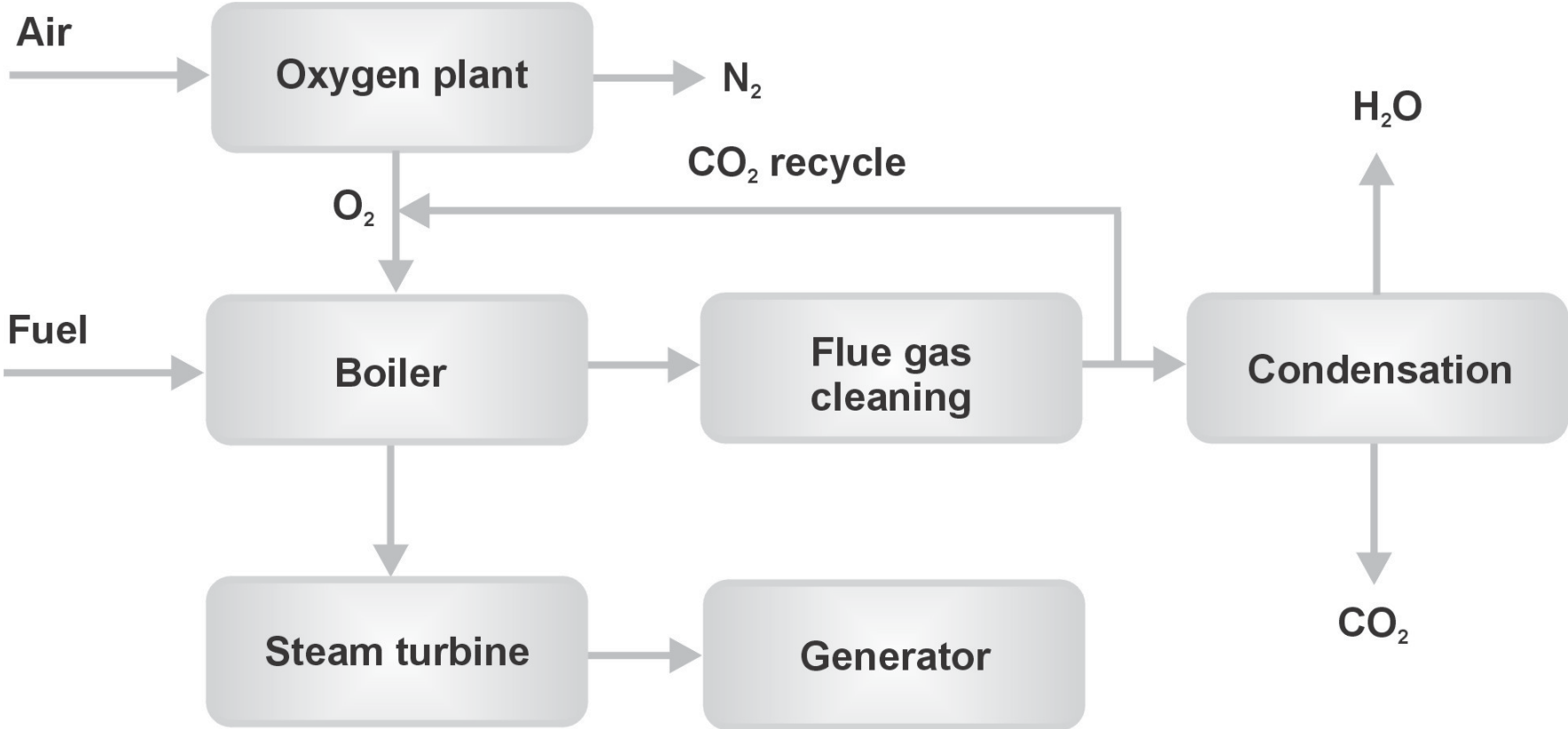
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INTRODUCTION



1. Increasing production of sewage sludge in Poland creates environmental problems with its disposal.
2. New legislation introduced in January 2016 limits landfilling of sewage sludge.
3. Thermal treatment of sludge becomes an attractive option because:
 - It significantly reduces the volume of waste material
 - It minimizes odour and destroys toxic compounds
 - It allows energy recovery from the sludge (LHV > 6.0 MJ/kg)
4. Problems associated with sewage sludge incinerators include:
 - Emission of greenhouse gases
 - Ash disposal
5. Oxy-fuel combustion technology is the most promising option for the reduction of greenhouse gas emissions from combustion of fossil fuels and incineration of biological and industrial waste materials.

OXY-FUEL COMBUSTION PROCESS



OBJECTIVE AND SCOPE OF INVESTIGATION



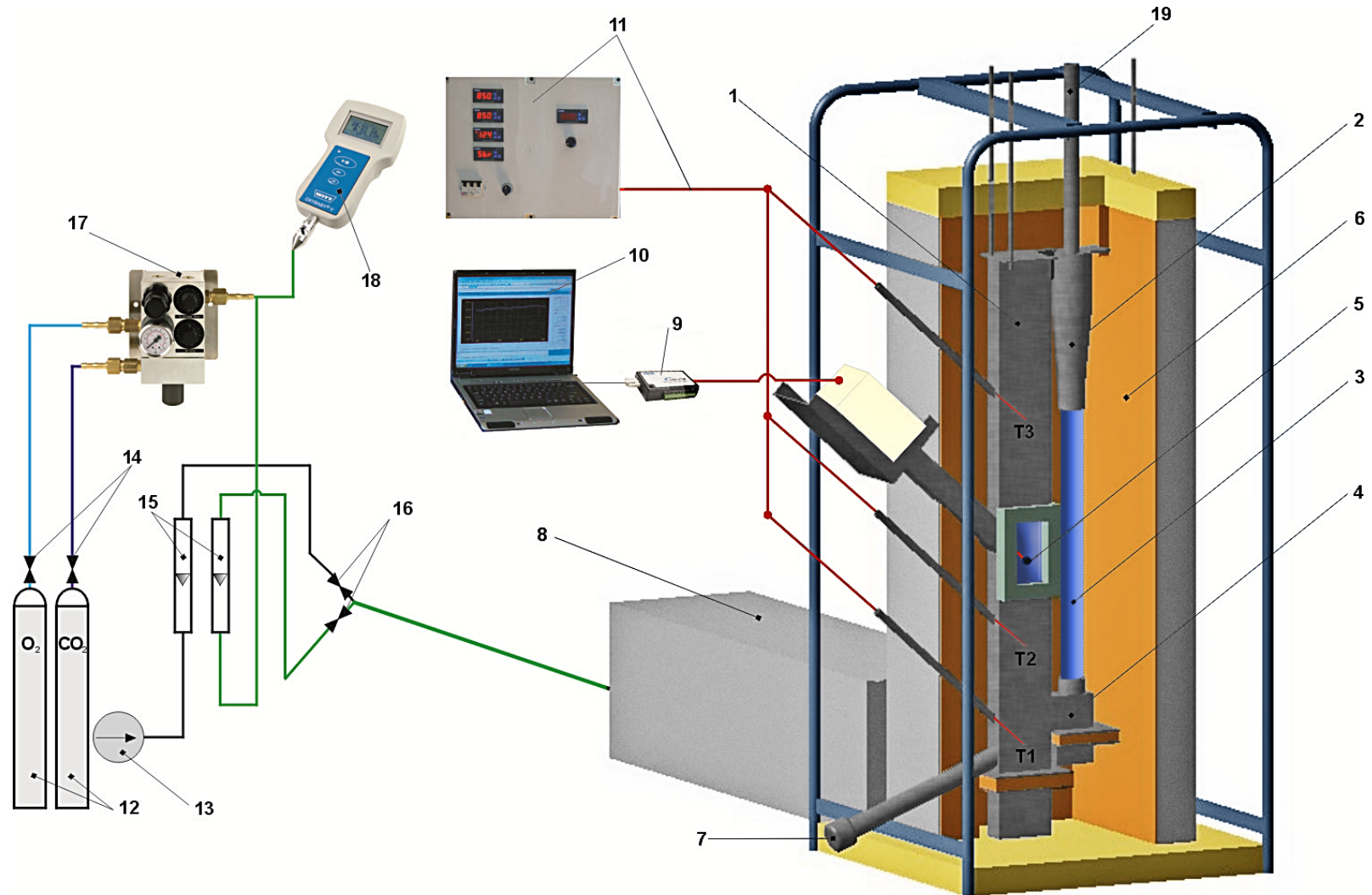
Objective: To determine the influence of fuel characteristic and oxidizing atmosphere on emissions of pollutants during combustion in a bench-scale circulating fluidized-bed reactor.

Scope:

1. Analyses of tested fuels,
2. Measurements of NO, N₂O and SO₂ during combustion in air and in O₂+CO₂ mixtures (oxy-fuel combustion),
3. Analysis of collected data.

EXPERIMENTAL APPARATUS

- 1 – combustion chamber
- 2 – cyclone
- 3 – downcomer
- 4 – loop seal
- 5 – fuel particles
- 6 – insulation
- 7 – drain valve
- 8 - preheater
- 9 – card
- 10 - computer
- 11 – temperature control system
- 12 – gas cylinders
- 13 – air compressor
- 14 - pressure regulators
- 15 – rotameters
- 16 – valves
- 17 – mixer
- 18 – gas analyser
- 19 – ventilation duct



T1–T3 – S-type thermocouples

EXPERIMENTAL CONDITIONS



Temperature: 850°C

Pressure: ambient

Bed material: silica sand, $d_{50} \sim 200 \mu\text{m}$

Fuels: dry sewage sludge, wooden biomass (willow), bituminous coal

Fuel sample mass: 0.5 g

Composition of fluidizing gas:

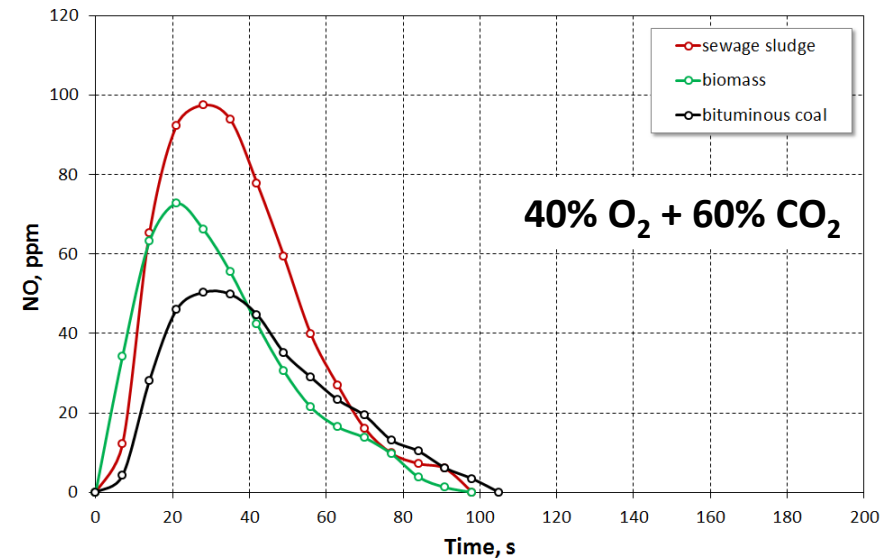
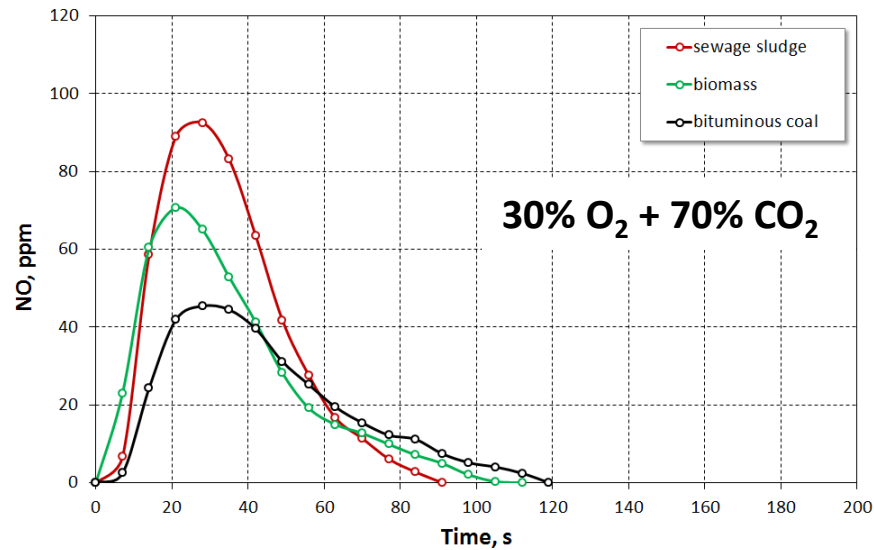
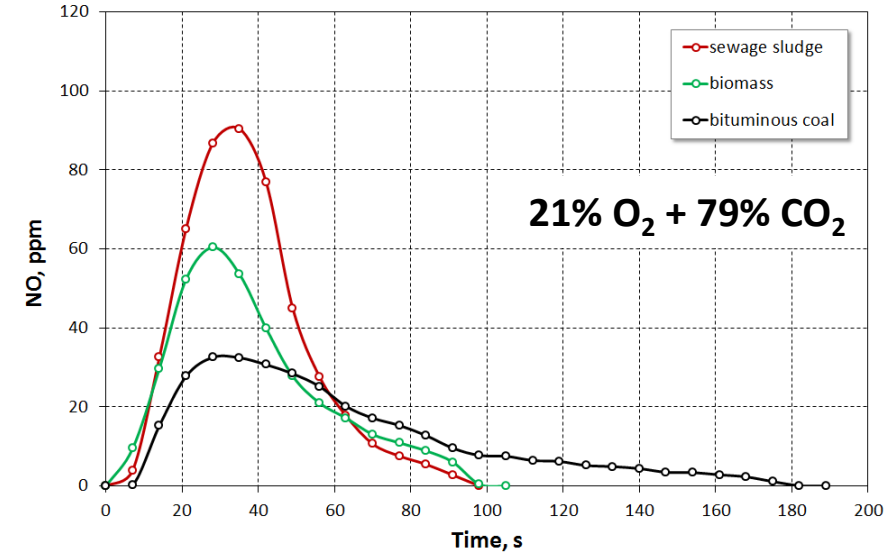
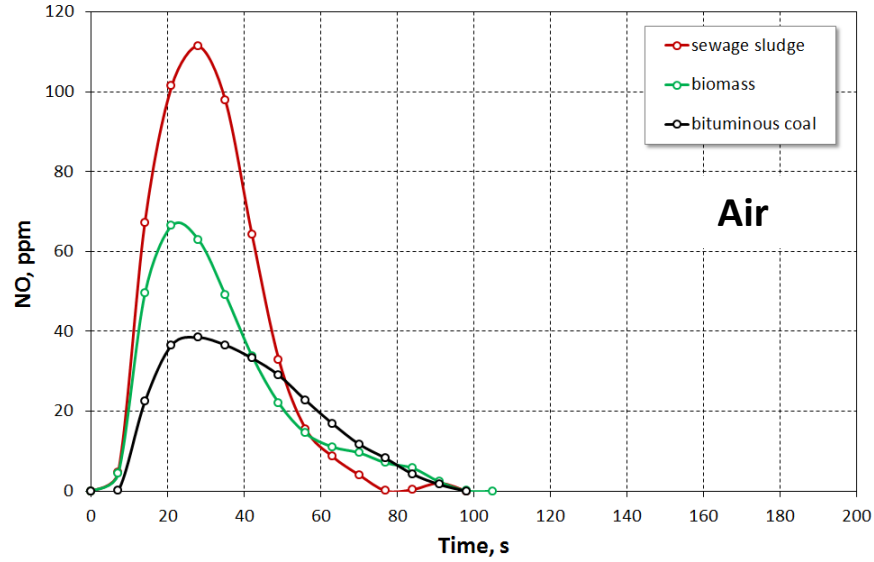
- 1. Air (base case)**
- 2. 21% O₂ and 79% CO₂**
- 3. 30% O₂ and 70% CO₂**
- 4. 40% O₂ and 60% CO₂**

ANALYSES OF TESTED FUELS



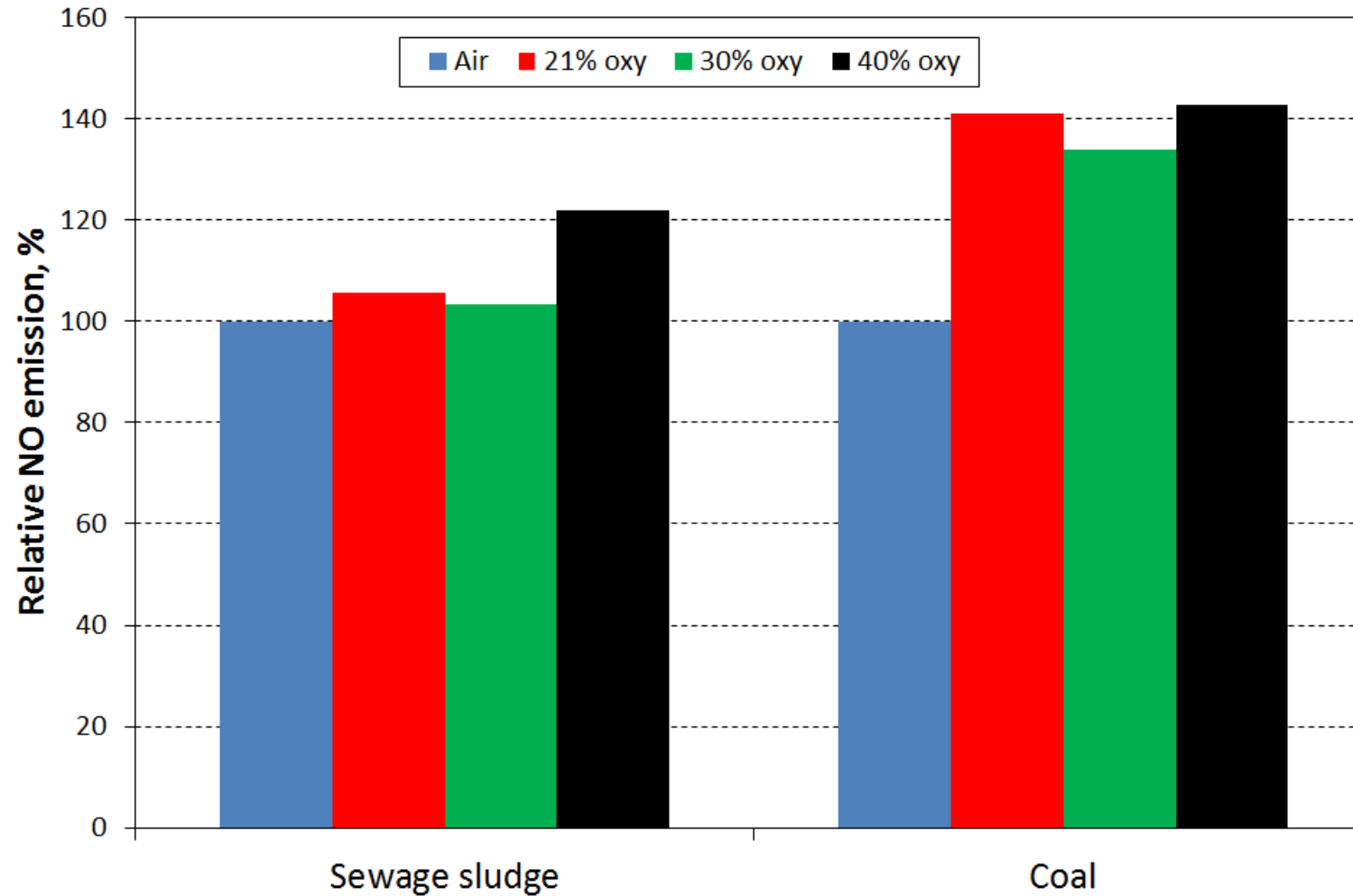
Fuel	Sewage sludge	Basket willow	Bituminous coal
Proximate analysis (air-dry basis)			
Moisture (M), %	4.9	6.9	8.7
Ash (A), %	36.4	1.4	18.9
Volatile matter (VM), %	51.4	76.3	26.8
Fixed carbon (FC), % (by difference)	7.3	15.4	45.6
Higher heating value (HHV), MJ/kg	13.55	18.20	22.75
Ultimate analysis (dry, ash-free basis)			
Carbon (C), %	52.49	49.59	73.30
Hydrogen (H), %	6.69	5.99	4.30
Sulphur (S), %	2.46	0.03	2.29
Nitrogen (N), %	7.27	0.33	1.10
Oxygen (O), % (by difference)	31.09	44.06	19.01

RESULTS – NO EMISSIONS

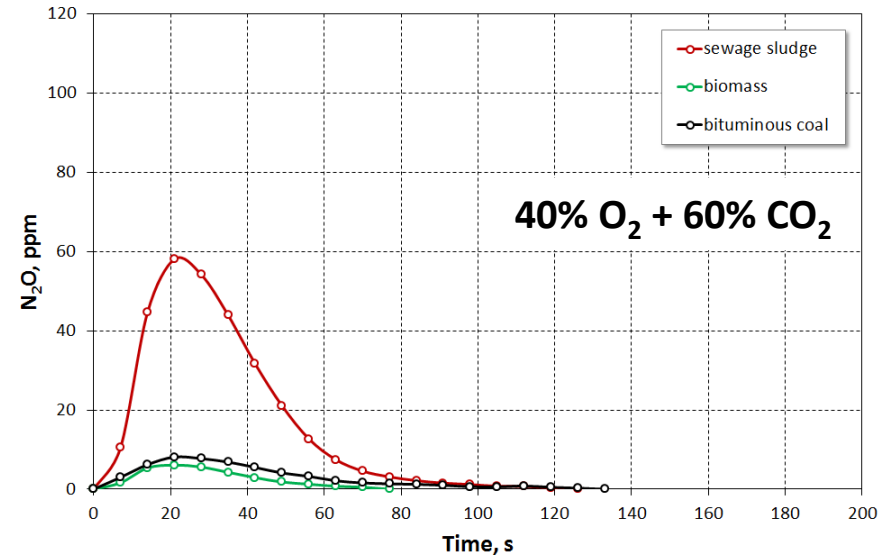
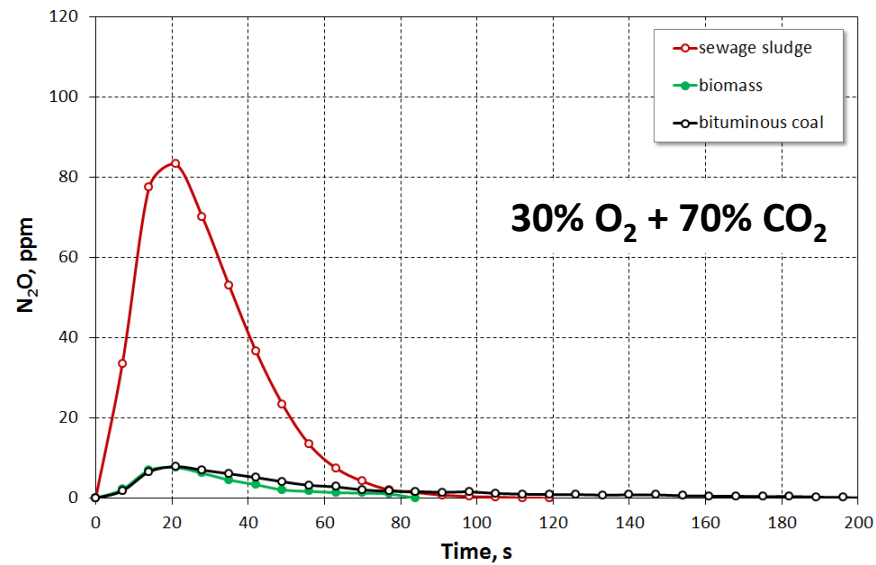
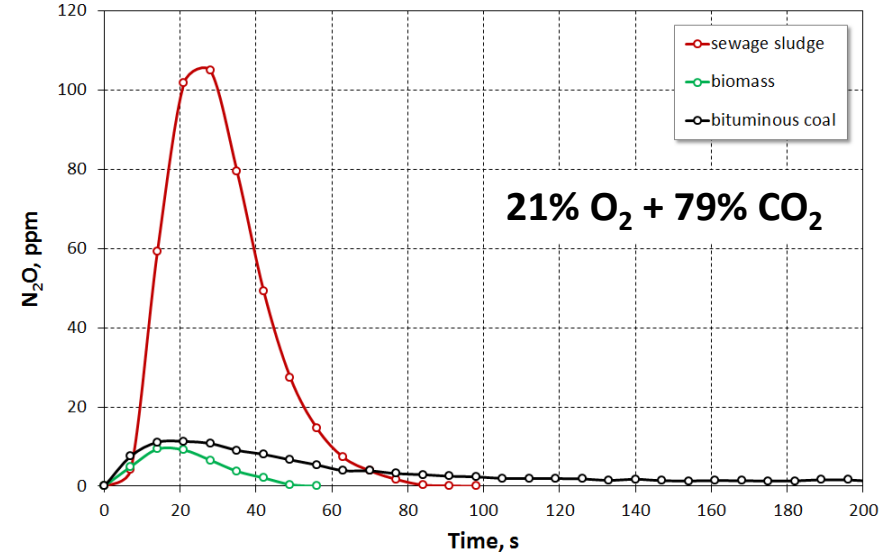
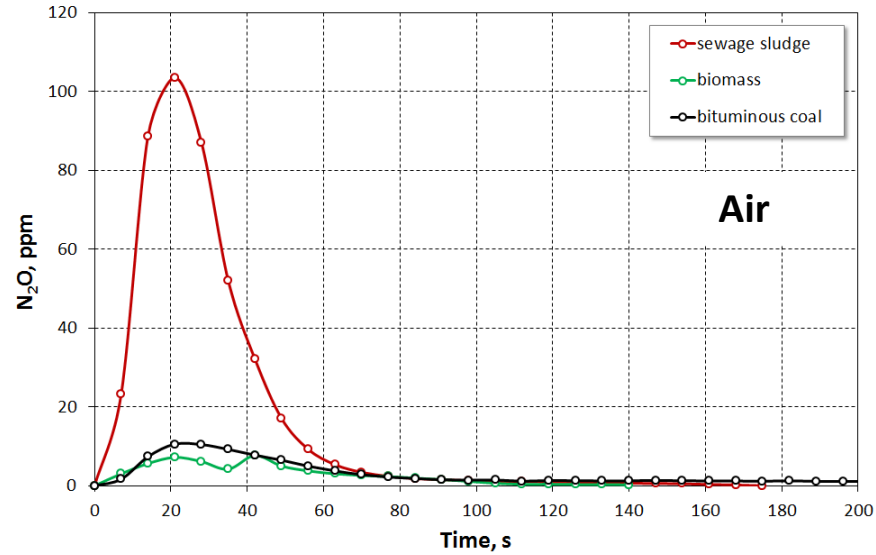


RESULTS – RELATIVE NO EMISSIONS

Influence of oxidizing atmosphere on NO emissions

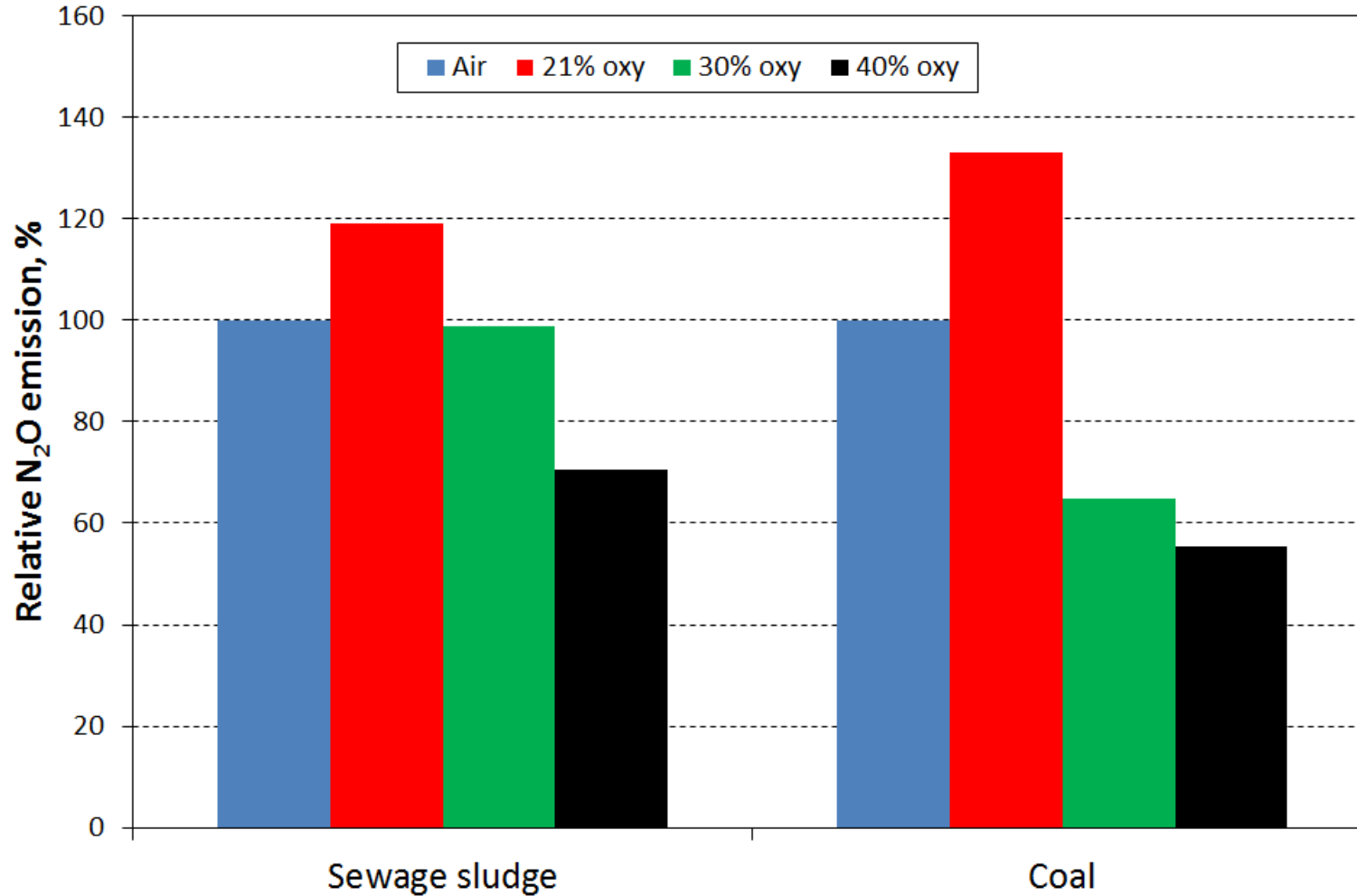


RESULTS – N₂O EMISSIONS



RESULTS – RELATIVE N₂O EMISSIONS

Influence of oxidizing atmosphere on N₂O emissions



RESULTS – CONVERSION OF FUEL-N TO NO_x



Conversion of fuel-N to NO, %

	Air	21% O ₂ and 79% CO ₂	30% O ₂ and 70% CO ₂	40% O ₂ and 60% CO ₂
Sewage sludge	9.3	9.9	9.6	11.3
Coal	25.6	36.1	34.2	36.5

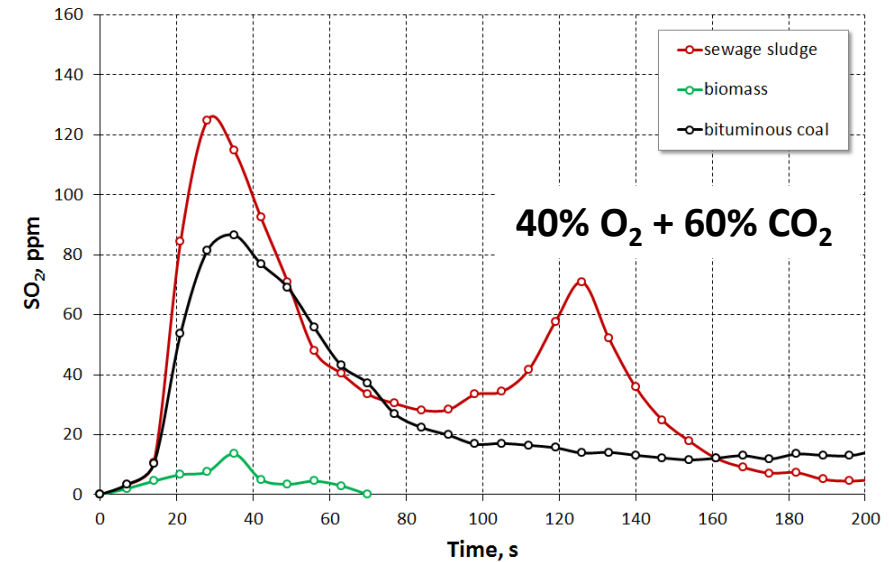
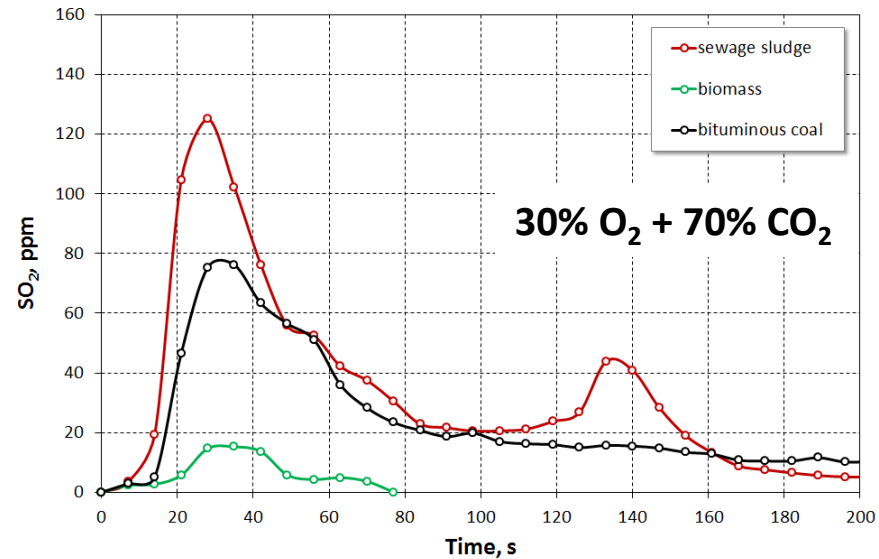
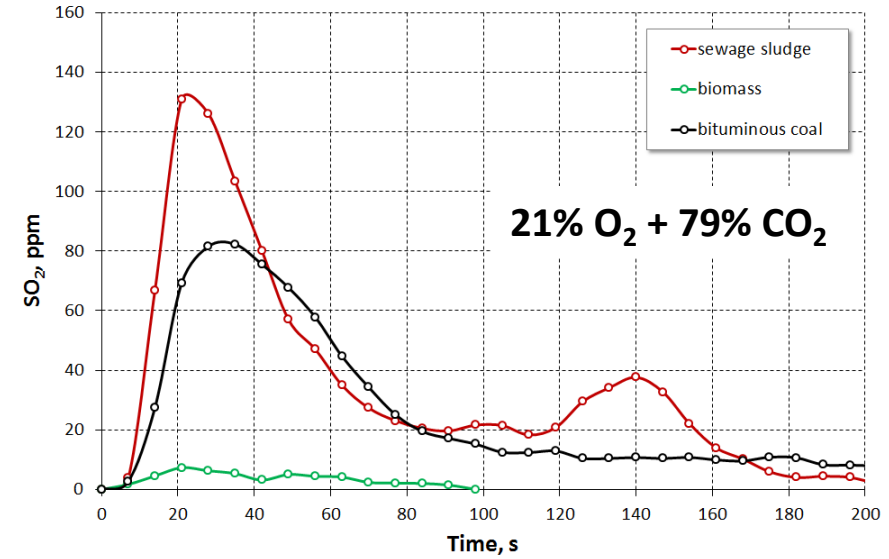
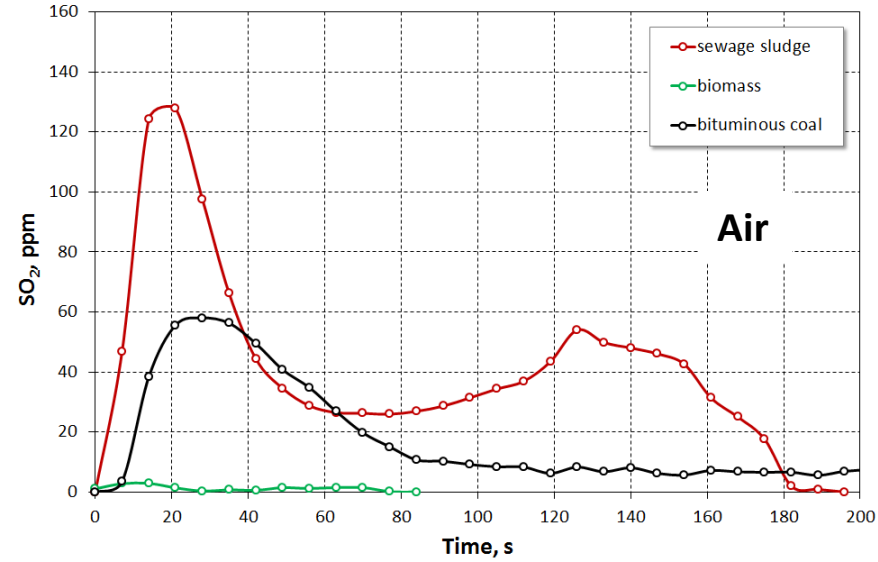
Conversion of fuel-N to N₂O, %

	Air	21% O ₂ and 79% CO ₂	30% O ₂ and 70% CO ₂	40% O ₂ and 60% CO ₂
Sewage sludge	15.9	18.9	15.7	11.2
Coal	19.9	26.2	12.9	11.0

Total conversion of fuel-N to NO_x, %

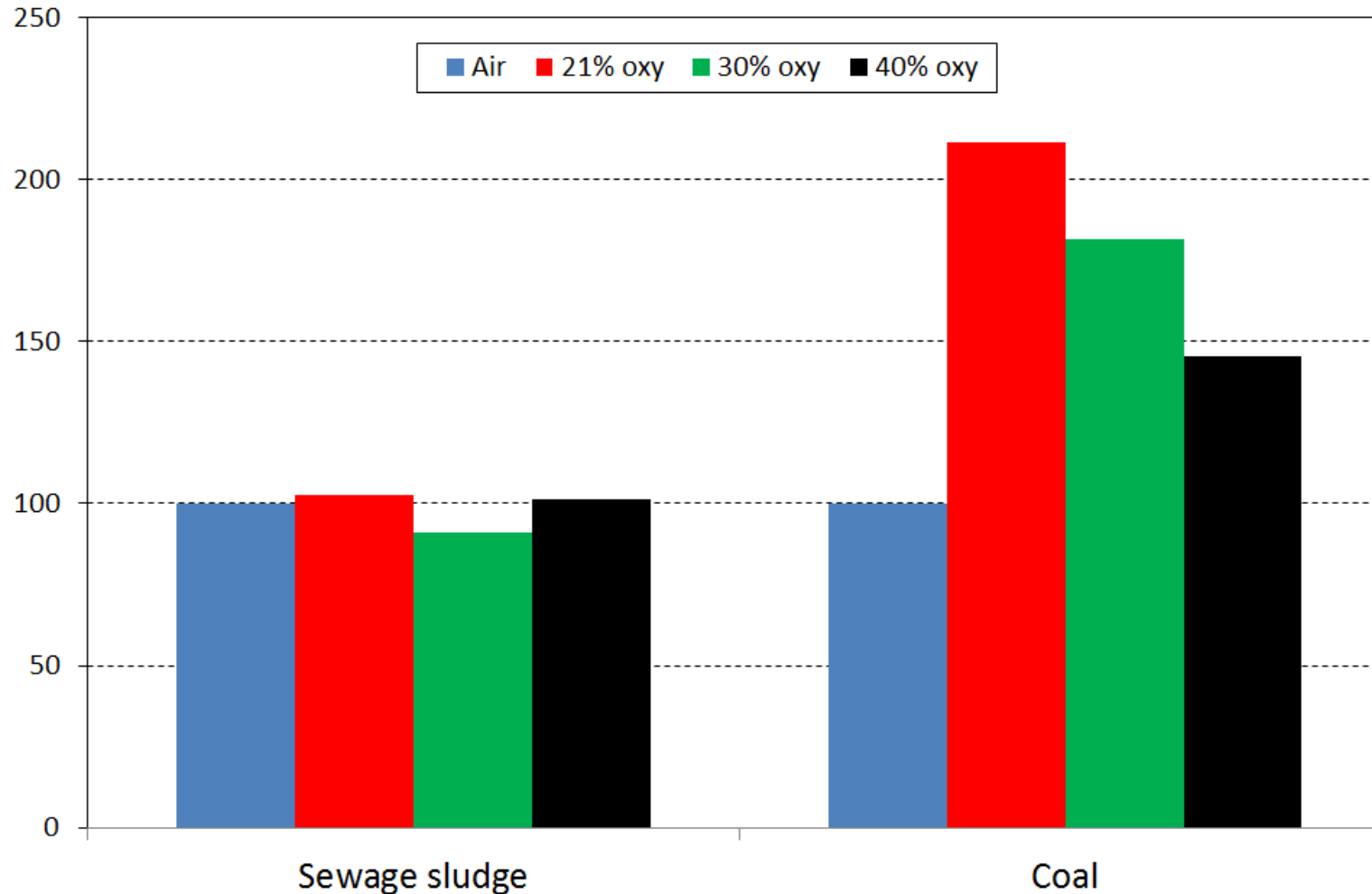
	Air	21% O ₂ and 79% CO ₂	30% O ₂ and 70% CO ₂	40% O ₂ and 60% CO ₂
Sewage sludge	25.2	28.8	25.4	22.6
Coal	45.5	62.6	47.2	47.6

RESULTS – SO₂ EMISSIONS



RESULTS – RELATIVE SO₂ EMISSIONS

Influence of oxidizing atmosphere on SO₂ emissions



CONCLUSIONS



1. Dry sewage sludge contains much more volatile matter, ash and nitrogen than the reference coal.
2. Instantaneous and average emissions of NO, N₂O and SO₂ for the combustion of sewage sludge in all atmospheres were much higher than those for the combustion of reference coal.
3. The SO₂ concentration profile for sewage sludge has a bimodal distribution which may suggest that SO₂ originated from both organic and inorganic sulphur sources.
4. Relative NO emissions for sewage sludge were insensitive to O₂ content up to 30% then they increased sharply. In the case of coal, combustion in O₂+CO₂ atmosphere caused sharp increase in NO emissions.
5. Relative emissions of N₂O for sewage sludge and coal have similar patterns. The highest emissions occurred for combustion in the 21% O₂ + 79% CO₂ mixture, then they decreased with increasing O₂ content.
6. SO₂ emissions for sewage sludge were insensitive to the composition of oxidizing atmosphere. In the case of coal, the highest emissions occurred for combustion in the 21% O₂ + 79% CO₂ mixture, then they decreased with increasing O₂ content.