

Thermochemical Energy Storage Employing Fluidized Bed Technology: Experimental Investigations with $\text{CaO}/\text{Ca}(\text{OH})_2$ on a 21kWh Reactor

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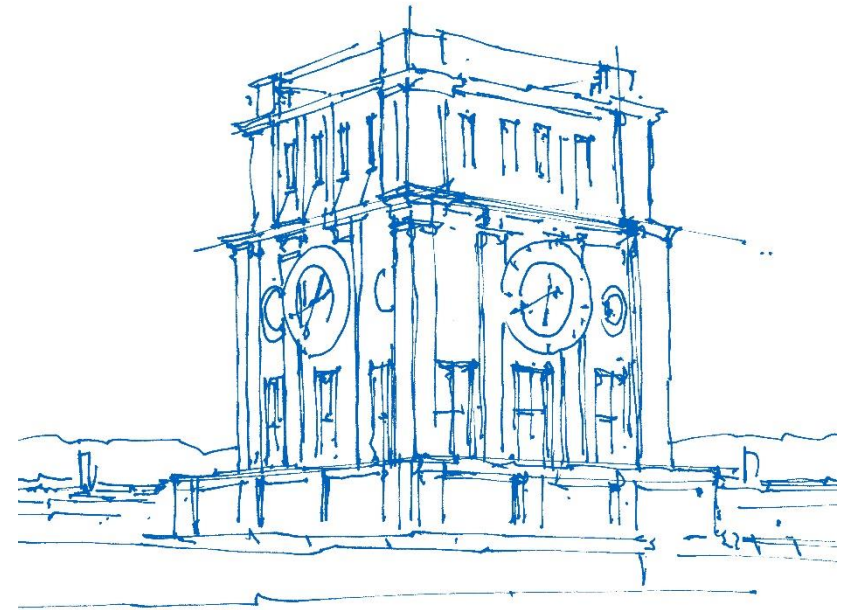
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Agenda

**Introduction:**

Thermochemical Energy Storage Systems

**Preliminary Experiments:**

Fluidisation Test Rig, Lab Scale Reactor, TGA

**Motivation & Open Questions:**

Heat Transfer and Particle Stability

**The Fluidized Bed Reactor FluBESToR:**

Setup and Experimental Procedure

**Experimental Data**

Thermochemical Energy Storage (TES)

Background & Potential

- First experimental investigations by Samms & Evans in 1967
- Water-based gas-solid-reactions advantageous
 → e.g. $\text{CaO(s)} + \text{H}_2\text{O(g)} \leftrightarrow \text{Ca(OH)}_2\text{(s)} + 104 \text{ kJ/mol}$
- Advantages:
 - Cheap storage material
 - No losses over time
 - High energy density
- Status Quo: Fixed bed and moving bed reactors [1]
- Current research: Moving bed [2] & fixed bed reactors [3]
- Advantages of fluidized bed reactors for TES:
 - Improved mass- and heat transfer
 - superior mixing properties
 - Simple scalability & compact design
 - Continuous operation allows decoupling of reactor volume and capacity

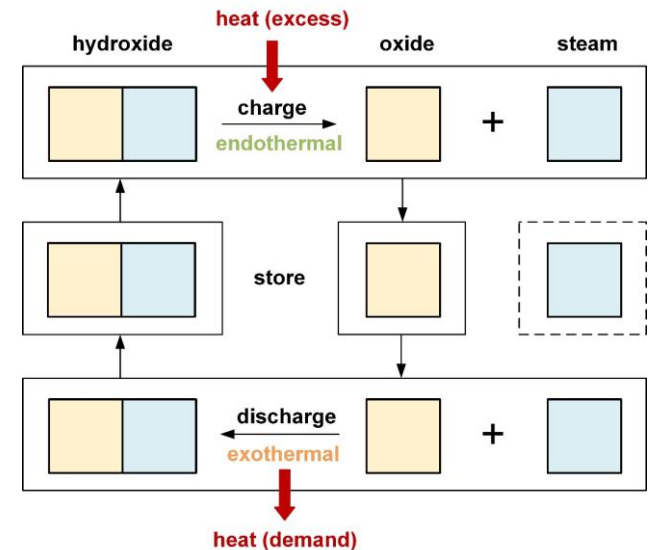


Illustration of the principle of thermochemical energy storage

Preliminary Experiments

Fluidization test rig with heat transfer probe

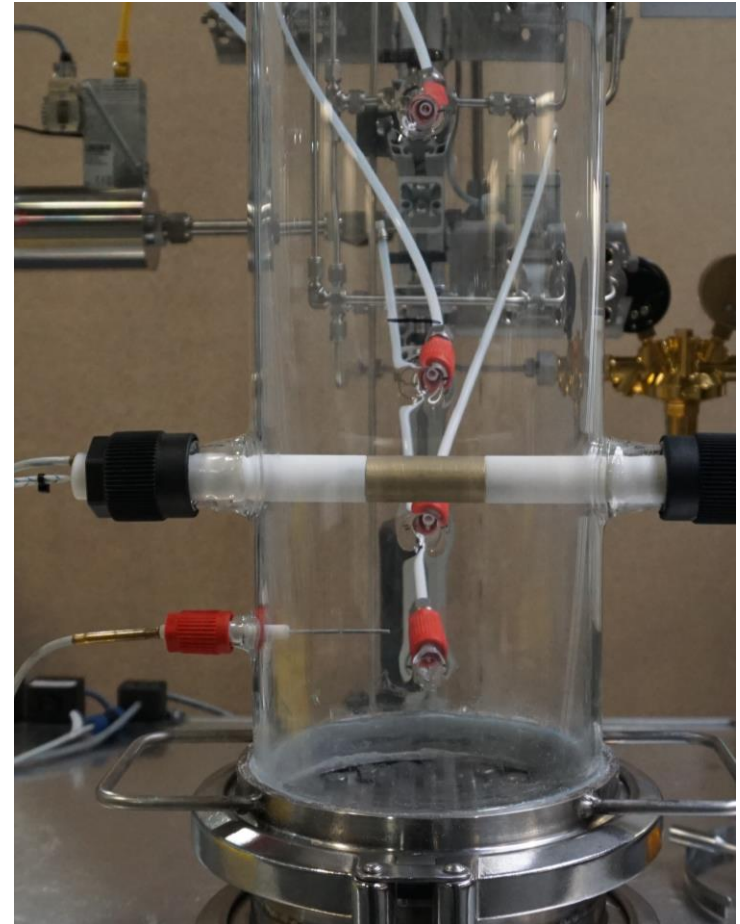
- Cylindrical columns (glass) with ID of 140 mm
- Dehumidified Nitrogen at ambient temperature as fluidization agent
- Overall perimeter heat transfer probe (OP-HTP), cp. [4,5]

Lab-scale fixed bed reactor

- Nitrogen and steam supply (direct evaporator)
- Radial flow trough porous material container
- Approx. 300 ml of storage material
- Temperature and pressure measurement

TGA unit

- Investigation of reaction kinetics
- Characterization and improvement of materials



Motivation & Open Questions

Heat-Transfer and Particle Stability in the Fluidized Bed

Heat Transfer

- Has been identified as limiting factor for overall process [6]
- Successful measurements in fluidized bed cold model (ambient conditions)

Cycle Stability

- Cycle stability so far not or hardly proven in literature
- Comminution processes in the process through three aspects:
 - Induced by chemical reaction
 - Induced by thermal stress
 - Mechanical forces in fluidized bed

Combination

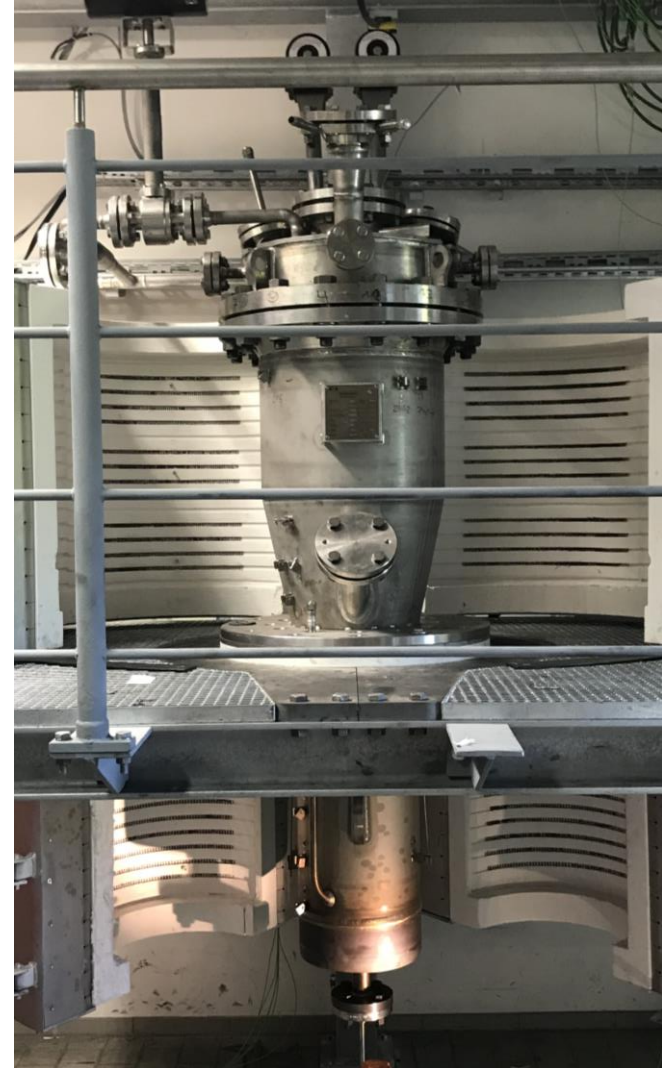
- Small quantity of fine materials leads to improvement of heat transfer temperature (higher collision factor → lower contact time)
- Too high a proportion of fine materials results in deterioration of heat transfer temperature (defluidisation)

Conclusion

- Definition of optimum operating point with specific fine fraction
- Particle stability for long term/high cycle experiments
- Estimation of heat transfer coefficients under operating conditions

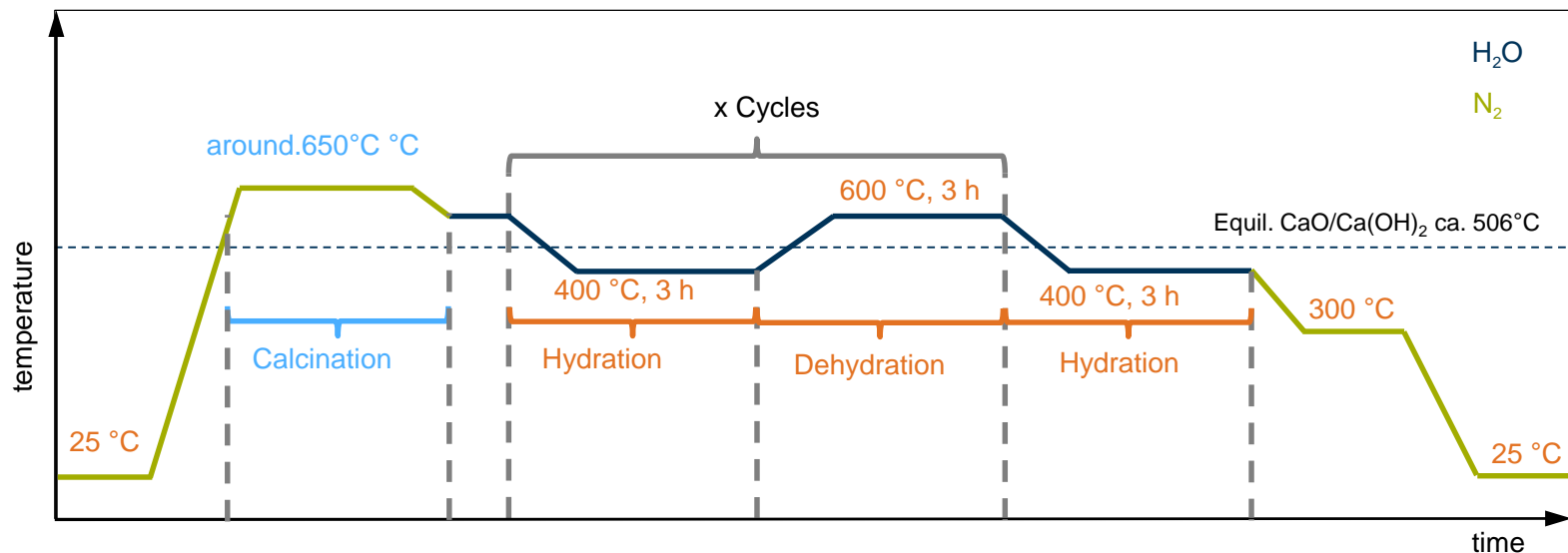
FluBEStoR

- Mounting plate with insulation ring
- Reactor torso
 - Temperature measurements
 - Gasinlet, gasbox and gas distributor plate
 - Flange for supplying bulk material
- Reactor-head
 - Sintered metal filter cartridges
 - Gasoutlet
 - Suction lance for solid sampling
 - Flange for material input in continuous operation mode
- Cooling coil (4,4 kW)
- Radiation furnace
 - 14 kW top
 - 40 kW bottom



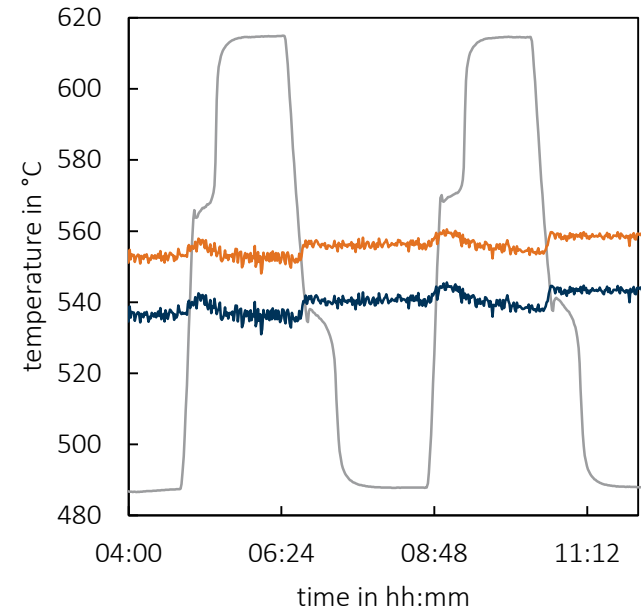
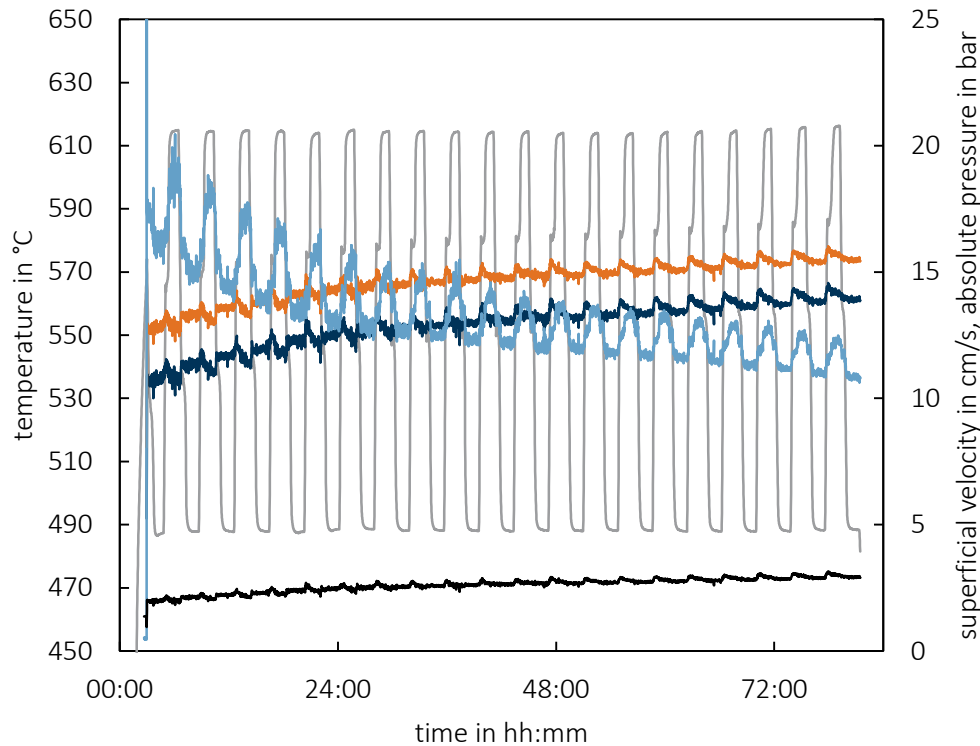
Exemplary FluBEStoR-Experiment

- “In-house” Calcination of CaCO_3 at around 650°C in pure nitrogen \rightarrow stable CaO
- One “Cycle” consists of complete hydration and dehydration
- Conversion close to chemical equilibrium is “gentle”
- Reaction itself is limited by heat input and -output
- Material analytics: Particle size analysis, TGA/ignition loss, scanning electron microscopy, BET, particle- & bulk density



Experimental Data

20.5-cycle experiment on lab scale reactor



— T Reaktor — Thyd — Tdehyd — superficial velocity — p Reaktor

Thank you for your attention!

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For additional literature on reactor see:

Wuerth M, Becker M, Ostermeier P, Gleis S, Spliethoff H. Development of a Continuous Fluidized Bed Reactor for Thermochemical Energy Storage Application. ASME. J. Energy Resour. Technol. 2019;141(7):070710-070710-6. doi:10.1115/1.4043629